

# Progress in Development of Silica Aerogel for Particle and Nuclear Physics Experiments at J-PARC

Makoto Tabata, Hideyuki Kawai

**Chiba University, Japan** 

The 2<sup>nd</sup> International Symposium on Science at J-PARC Unlocking the Mysteries of Life, Matter and the Universe

July 12–15, 2014 Tsukuba, Japan

P-404

#### Highlights

We are developing silica aerogels for use in particle and nuclear physics experiments at J-PARC.
Modernized conventional method allows us to produce highly transparent aerogels at n = 1.03–1.05.
Pin-drying method enables us to generate ultrahigh-refractive-index aerogels up to n = 1.26.

### References

• M. Tabata et al., Nucl. Instrum. Methods A 623 (2010) 339.

- M. Tabata et al., Nucl. Instrum. Methods A 668 (2012) 64.
- M. Tabata et al., Physics Procedia 37 (2012) 642.

### Introduction

• Silica aerogel

- Tunable refractive index, n (determined by silica–air volume ratio)
  - n = 1.003–1.11 (by our conventional KEK method)
  - n = 1.05–1.26 (by novel pin-drying method)
  - $\rightarrow$  density,  $\rho = 0.01 1.0 \text{ g/cm}^3$

## **Our Conventional (KEK) Production Method**

- KEK method: our standard method for producing hydrophobic aerogels
  - Developed for producing aerogels with n = 1.01 1.03 in 1990s
  - For use in the aerogel Cherenkov counters (ACCs) of Belle experiment at the KEKB collider
- Using ethanol or methanol as solvent in the wet-gel synthesis
  Down to n = 1.003 (for use as cosmic dust intact capture medium)
  Modernized conventional method

- Long transmission length,  $\Lambda_{T}$ 
  - e.g.,  $\Lambda_{T} = 40$  mm at n = 1.05 ( $\lambda = 400$  nm)
- Hydrophobic material (by our production method)
  Suppressing age-related degradation caused by moisture absorption
  Large volume, V

• V<sub>max</sub> = 18 × 18 × 2 cm<sup>3</sup> (n ~ 1.05), 15 × 15 × 3 cm<sup>3</sup> (n ~ 1.03)
• To be used in several particle and nuclear physics experiments at J-PARC

• As a Cherenkov radiator (particle identification)

n = 1.009

• Threshold Cherenkov counters

• e.g., E03 (n = 1.12), E07 (n ~ 1.17), E14 (KOTO, n = 1.03), E36 (TREK, n = 1.08), E42 (n = 1.05)

Ring imaging Cherenkov counter
e.g., E50 (n = 1.03–1.05)

As an ultra-cold muon source (muonium production target)

• E34 (muon g–2/EDM,  $\rho = 0.03$  g/cm<sup>3</sup>, n = 1.008)

Nanostructure of silica aerogel n = 1.003

- Developed for producing aerogels with n = 1.045–1.06 in 2000s
  For use in the aerogel ring imaging Cherenkov (A-RICH) detector of Belle II experiment scheduled at the SuperKEKB collider
  Using N,N-dimethylformamide (DMF) or mixture of DMF and methanol
  Up to n = 1.11
- Production procedure (~ 1 month)

  Wet-gel synthesis (molding)
  Aging and mold detachment
  Hydrophobic treatment
  Solvent replacement (using ethanol)
  Supercritical drying with an autoclave (using carbon dioxide or ethanol)





Scanning electron microscope image



### **Pin-drying Method**

**Dropped** water

- Recently developed for producing aerogels with ultrahigh refractive index (n > 1.10)
- Production procedure
  - 1. Wet-gel synthesis
  - 2. Aging and mold detachment
     ← 3. Pin-drying process
  - 4. Hydrophobic treatment5. Solvent replacement6. Supercritical drying





1. **N** 

1.15

Refractive index



Before hydrophobic treatment, to increase silica density, the wet gel is shrunk by partial drying (solvent evaporation) in a semi-sealed container punctured with pin holes to suppress cracking the wet gel.
More transparent aerogels can be obtained after supercritical drying. • Using the modernized conventional method:  $\Lambda_{T} = 40-70 \text{ mm}$ at n = 1.03-1.05• Using the pin-drying method:  $\Lambda_{T} = 40-60 \text{ mm}$ at n = 1.05-1.075 $\Lambda_{T} > 20 \text{ mm}$ at n = 1.075-1.26

1.00

1.05

#### n = 1.06, thickness = 1 cm

1.20

1.25

1.30

Classical aerogel  $\Lambda_{\rm T} = 14$  mm

Pin-dried aerogel here!  $\Lambda_{\rm T}$  = 53 mm